

Application of high-strength biodegradable polyurethanes containing different ratios of biobased isomannide and poly(ϵ -caprolactone) diol

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Dong-In Lim¹, Hyung-Seok Park¹, Jeong-Hui Park¹,
Jonathan C Knowles^{1,2} and Myoung-Seon Gong¹

Abstract

Biodegradable–biocompatible polyurethanes were prepared with fixed hexamethylene diisocyanate and varying ratios of isomannide and poly(ϵ -caprolactone) diol using a simple one-step polymerization without a catalyst. The polyurethane structures were confirmed by ¹H-nuclear magnetic resonance, Fourier transform infrared spectroscopy, and gel permeation chromatography. The glass transition temperatures were determined by thermal analysis to be between 25°C and 30°C. Degradation tests performed at 37°C in phosphate buffer produced mass losses of 5%–10% after 8 weeks. After 5 days of culture, using osteoblastic cells, the relative cell number on all the polyurethane films was only slightly lower than that of an optimized tissue culture plastic. These polymers offer significant promise with a simplistic synthesis and controlled degradation.

Keywords

Biodegradable, biobased monomer, polyurethane, biocompatible, isomannide

¹Department of Nanobiomedical Science & WCU Research Center, Graduate School, Dankook University, Cheonan, South Korea

²Department of Biomaterials and Tissue Engineering, UCL Eastman Dental Institute, London, UK

Corresponding author:

Myoung-Seon Gong, Department of Nanobiomedical Science & WCU Research Center, Graduate School, Dankook University, Cheonan, 330-714, South Korea.

Email: msgong@dankook.ac.kr

Introduction

Tissue engineering is the future medical treatment for reconstructing defective tissues over the last decade.¹ Tissue regeneration is a relatively simple tissue engineering treatment compared with other treatments, such as in vitro tissue reconstruction; for example, in periodontitis, new tissue was reconstructed using a membrane to prevent other tissues from invading the affected areas.² Recently, tissue regeneration, and the biomaterials used in tissue regeneration, has undergone a significant expansion both academically and commercially.³

The major materials used in tissue regeneration are biodegradable polymers. Biodegradable polymers, with good biocompatibility, have significant potential in various tissue engineering and drug delivery fields.⁴ Virtually, all tissue engineering devices are implanted in a mechanically active site in the body; thus, these implants are subjected to dynamic stress; therefore, it is very important that the polymers do not physically degrade, as their by-products can cause inflammatory effect to the surrounding tissues. Several families of biodegradable polymers have been developed, such as polyurethanes, poly(aliphatic esters), and poly(phosphate esters).⁵ These polymers have dominated the field as they are relatively cheap and they have Food and Drug Administration approval for clinical use.

Polyurethanes are a class of biodegradable polymers that have been applied as tissue engineering scaffolds as they show low in vitro and in vivo cytotoxicity.⁶ Polyurethanes created by reacting dialcohol with diisocyanate result in the highly hydrogen-bonded urethane linkages with good physical and mechanical properties and with the haemocompatibility and flexibility needed for use in biomedical applications.⁶ In fact, many non-biodegradable urethanes have been utilized in blood contact applications such as breast implants, aortic grafts, bone adhesives, dialysis membranes, and heart valves.⁶ Biodegradable polyurethanes can be obtained using biodegradable soft segments and isomannide hard segments. In the biodegradable soft segment, polyurethanes, such as those containing poly(ϵ -caprolactone) (PCL), have been obtained⁷ along with poly(ethylene oxide) (PEO) and poly(L-lactide) PLA.⁸ In the biodegradable hard segments, the diisocyanate and the chain extender can be designed from a variety of biologically relevant molecules.

Several possibilities were considered to produce biodegradable polyurethanes. Biodegradable hard segments have been obtained with aliphatic diisocyanate, methylene lysine diisocyanate, and hexamethylene diisocyanate (HDI). Good examples of such raw materials are those produced from glucose as the natural feedstock.⁹ Isomannide is a promising material for biobased monomers as it is nontoxic and has appropriate chirality and bicyclic ring structure, which imparts rigidity.¹⁰ Isomannide has been used in various polymer systems such as polyester, polyether, and polyurethane.

In this study a polyurethane series was made with isomannide, PCL diol, and HDI compounds to satisfy the demands of tissue regeneration technology. Aliphatic diisocyanates are usually chosen to synthesize biocompatible and biodegradable polyurethanes because their degradation product is nontoxic. Polyester diols, such as PCL diol, are also chosen as they hydrolytically degrade to caproic acid. Isomannide was chosen as the dialcohol monomer for its chiral, rigid, and nontoxic properties. The typically polyurethanes are made in a two-step polymerization.¹¹ However, a bulk, one-pot synthetic method without catalyst, was employed in this study.

Experimental section

Materials and instruments

Isomannide (98%) and HDI (99%) were purchased from Sigma–Aldrich (St Louis, MO, USA). PCL diol 2000 (CAPA 2201A; $M_w = 2000$) was purchased from Kangnam Chemical Co., Ltd (Seoul, South Korea), and phosphate-buffered saline (approximately pH 7.3) was

Table 1. Polyurethanes with different composition contents and their yields.

Polyurethane	HDI	Isomannide	PCL diol (M _w = 2 kDa/mol)	Yield (%)
	Mole ratio			
PU 1	8	3	5	87
PU 2	8	4	4	78
PU 3	8	5	3	71
PU 4	8	6	2	59

HDI: hexamethylene diisocyanate; PCL: poly(ϵ -caprolactone).

obtained from Oxoid Ltd (Hampshire, England). All chemicals were used without further purification.

Preparation of the polyurethanes

The monomers were mixed together and allowed to react.¹¹ Synthesis was carried out in a 250 mL round-bottomed four-neck flask with reactant stoichiometry of 1:1 with diisocyanate:diol (PCL diol and isomannide) under a dry nitrogen atmosphere. A nitrogen flushed four-neck flask equipped with a mechanical stirrer, thermometer, and condenser was charged with 12.5 mmol of PCL diol (25 g) and 12.5 mmol of isomannide (41.83 g) (reactant stoichiometry and precursor weights are given as an example for polyurethane (PU) 1; Table 1). The flask was maintained at 80°C and stirred under a nitrogen atmosphere. After the solid contents were melted completely, 25 mmol of HDI (4.21 g) was added and stirred for 5 min. The sample was allowed to stand in a Teflon dish to react for 12 h at 150°C. The synthesized polyurethanes were dissolved in *N,N*-dimethylformamide (DMF). After the resulting solution was poured into a large amount of isopropyl alcohol, a polymer precipitate was formed and filtered. The product was dried at 40°C for 72 h under vacuum and stored in a desiccator. PU 2, PU 3, and PU 4 were prepared by similar procedures. Polymer yields of 87%, 78%, 71%, and 59% were obtained. The formulation used is described in Table 1.

Polymer characterization

¹H-nuclear magnetic resonance (NMR) spectrum was recorded for the synthesized polyurethanes with a Bruker Avance 600 spectrometer at 600 MHz (Bruker Co., Germany) and performed at ambient temperature with 5% (w/v) polymer solution in CDCl₃. Fourier transform infrared (FT-IR) spectra were obtained at room temperature using a Bio-Rad Excaliber TS-3000MX (Bio-Rad, Tokyo, Japan) in the range of 800–4000 cm⁻¹. A 2.5% solution of polymer in chloroform was placed directly onto a KBr pellet (Sigma–Aldrich), and the chloroform was evaporated at 50°C under vacuum for 2 h. The spectra were also checked for evidence of residual solvent.

Molecular weight

The weight average (M_w) and number average (M_n) molecular weights of the polyurethanes were measured by gel permeation chromatography (GPC) using a Futecs NP-4000 instrument (Futecs, Seoul, South Korea) equipped with a model P-4000 pump, a model AT-4000 column oven, a GPC KF-804 column, and a Shodex (Shodex, Tokyo, Japan) R1-101 refractive index detector.

Tetrahydrofuran (THF) was used as the eluent at a flow rate of 1.0 mL min^{-1} , and a sample concentration of 2.5 mg mL^{-1} was used. Polystyrene ($M_w = 2000, 7000, 12,000, 65,000$, and $120,000$) was used as the standard.

Thermal analysis

Differential scanning calorimetry (DSC) data were recorded with DSC (SEIKO Exstar 7000, Tokyo, Japan) instrument. Specimens (approximately 10 mg) were sealed in a DSC aluminum pan before being placed in the calorimeter. The samples were cooled to -70°C and then heated to 250°C at a rate of $10^\circ\text{C min}^{-1}$ under a nitrogen atmosphere.

Mechanical properties

Following synthesis of the polymer, polymer films for mechanical testing and degradation testing were prepared by solvent casting in DMF at 10% polymer concentration into a polytetrafluoroethylene (PTFE) dish followed by air drying to produce films of 0.2 mm thickness. Tensile strength and elongation at break of the polyurethanes were measured on an Instron universal testing machine (Model 3344; Instron Engineering Corp., Canton, MA, USA) at a crosshead speed of 10 mm min^{-1} at room temperature. The specimens were prepared with a dog bone-shaped cutter. The thickness and width of the specimens were 0.2 and 5 mm, respectively. The length of the sample between the grips of the testing machine was 15 mm. Five specimen measurements were conducted for each polyurethane, and the results were averaged to obtain a mean and standard deviation.

Contact angle and surface energy

The wetting ability of the polymer surface was evaluated based on contact angle measurements using a Phoenix 300 contact angle equipment (SEO, Seoul, South Korea). Tests were carried out using the sessile drop mode with ultrapure water, which was used to represent polar characteristics. Approximately $5 \mu\text{L}$ droplets of the test liquid were placed on polymer specimens using a manual syringe. The drop profile was then recorded at an interval of 1 s for 1 min, and the measurements were carried out on triplicate samples. The contact angle values were evaluated using Phoenix 300 software. The polyurethane series was dried for 24 h prior to contact angle measurement. The calculation of the surface energy was conducted using the Girifalco–Good–Fowkes–Young method and Phoenix software.

In vitro degradation test

Polyurethane film ($\Phi = 12 \text{ mm}$) degradation was quantified by the changes in dry weight. The polyurethane samples were allowed to degrade for 2, 4, and 8 weeks. Dry films were weighed (w_0) and immersed in a conical tube containing 10 mL phosphate-buffered solution (approximately $\text{pH} = 7.3$). The degradation was conducted at $37^\circ\text{C} \pm 1.5^\circ\text{C}$ in a water bath (WB-6; DAIHAN Scientific Co., Ltd, Seoul, South Korea). Samples were taken at intervals, rinsed with water, dried in a vacuum oven for 2 days at 50°C , and weighed (w_t), after which they were discarded. The weight remaining was calculated as

$$\text{Weight remaining (\%)} = (w_0 - w_t / w_0) \times 100$$

Cell culture

The polyurethane films were sterilized by soaking in 50%, 70%, and 100% ethanol for 30 min prior to use, and then they were dried for 2 h. MC3T3-E1 cells were maintained in standard T75 tissue culture flasks in normal growth medium composed of alpha-modified minimum essential medium (α -MEM) (Invitrogen, Paisley, UK) supplemented with 10% fetal bovine serum (Gibco, Daejeon, South Korea) and 1% penicillin/streptomycin (Gibco). Prior to cell seeding, sections were cut from the synthesized polyurethane films, placed into 96-well plates, and seeded with 5×10^3 cells suspended in 1 mL normal growth medium and maintained at 37°C with 5% CO₂ for a subsequent time course analysis of cell number.

Cell proliferation

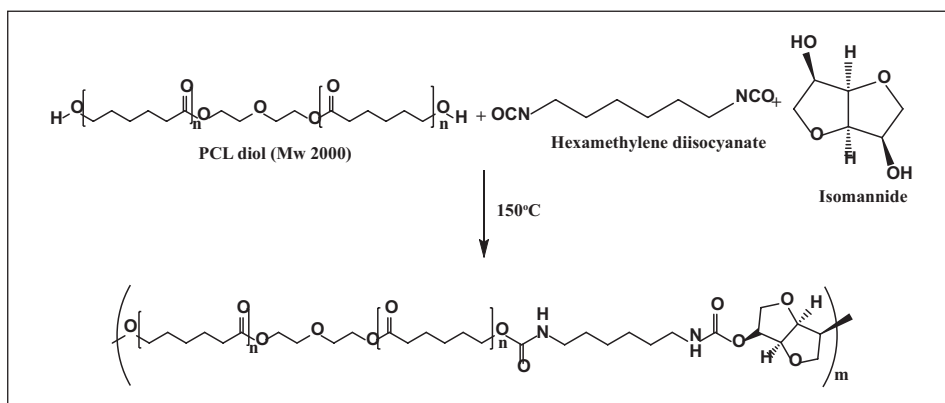
Cells were cultured on the polyurethane films ($n = 3$) (the films were held down in the culture plates with PTFE inserts) in 96-well plates for 1, 3, and 7 days. Cell proliferation was determined at these time periods using the cell counting kit-8 (CCK-8) (2-(methoxy-4-nitrophenyl)-3-(4-nitrophenyl)-5-(2,4-disulfophenyl)-2H-tetrazolium, monosodium salt) assay, according to the manufacturer's instructions (CCK-8; Dojindo Laboratories, Kumamoto, Japan). The cells were then incubated for 1, 3, and 7 days, and the cell numbers were measured using the CCK-8 reagent. The CCK-8 solution was freshly prepared in growth medium, and 200 μ L of CCK-8 reagent was added to each well, followed by incubation in a humidified atmosphere with 5% CO₂. Absorbance was measured at 450 nm using a spectrophotometer (Bio-Rad, Seoul, South Korea). A blank experiment to detect cell-free background absorbance was also performed in parallel. Results are expressed as relative CCK-8 activity compared to control conditions (cell-free background absorbance).

Results

A family of biocompatible and biodegradable polyurethanes was synthesized by a simple one-pot polymerization of PCL diol ($M_w = 2000$) and isomannide, as shown in Scheme 1. The reaction was carried out without a catalyst or solvent, which are usually required, in order to decrease potential toxicity.¹³ The chemical structures of the polyurethanes obtained were characterized by ¹H-NMR and FT-IR spectroscopy.

Figure 1 displays the ¹H-NMR spectrum of polyurethane in CDCl₃, in which all proton signals of isomannide, PCL diol 2000, and HDI segments were confirmed. The signals occurring from 3.5 to 5.2 were assigned to methylene protons of isomannide. In addition, the peak at 3.1 parts per million (ppm) was assigned to the amine proton of the urethane linkage.

In the FT-IR spectra, the complete disappearance of the isocyanate peaks in the polyurethane samples occurred at 2250 cm⁻¹ (Figure 2),¹⁴ indicating that no unreacted isocyanate groups were present in the polyurethane samples. After polymerization, the -OH peaks in the area around 3446–3558 and 3427 cm⁻¹ of PCL diol and isomannide, respectively, disappeared. A new peak at 1529 cm⁻¹ appeared, which was caused by the amide (N-H) bending vibration of the polyurethanes. The FT-IR spectrum confirmed that the polyurethane structures had a pronounced free amide group (-NH-) peak at 3322 cm⁻¹, and hydrogen-bonded NH stretching bands were seen at 1529 cm⁻¹,¹⁵ and a carbonyl group (C=O) stretching transmittance at 1721 cm⁻¹. Methylene group polymer chains were represented by peaks between 2938 and 2860 cm⁻¹ in the polyurethanes and PCL diol. C-H bending and C-O stretching vibration bands were seen in the area from 1500 to 600 cm⁻¹.



Scheme 1. Synthetic of the route and structure of the polyurethane series.

PCL: poly(ϵ -caprolactone).

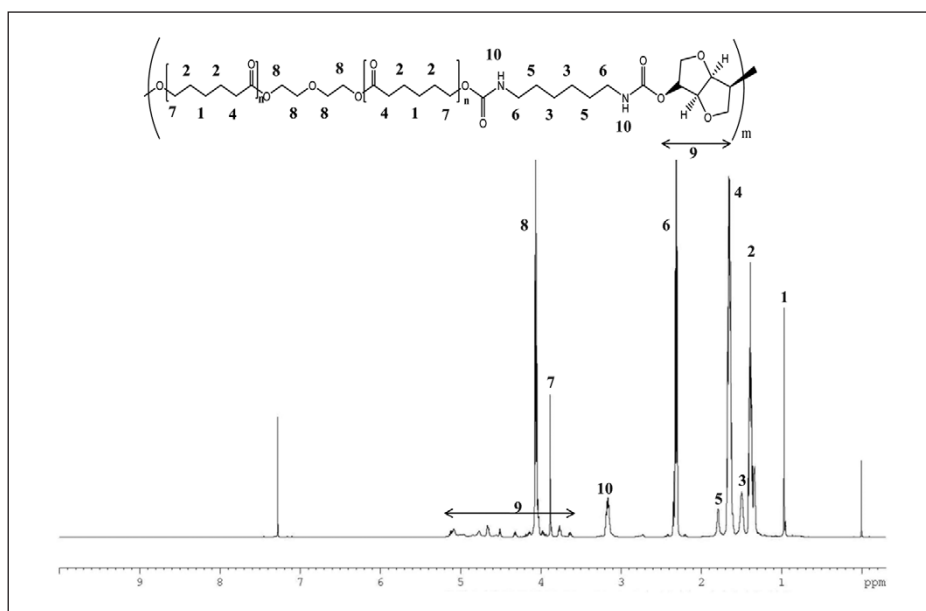


Figure 1. ^1H -nuclear magnetic resonance (NMR) spectra of polyurethane.

Molecular weights and thermal properties

The thermal properties and molecular weights of the polyurethanes are listed in Table 2. The weight average molecular weights (M_w) ranged from 16,892 to 145,070. Polyurethanes with the higher levels of isomannide yielded lower M_w compared to polyurethanes with low levels of isomannide may be due to the presence of a secondary hydroxyl group, which is lower in reactivity compared to the primary hydroxyl group¹² of the PCL diol. It is important to understand the thermal behavior of polymers developed for biomedical applications, as it determines the physical properties of the materials and the processability. The DSC data of the synthesized polyurethane series are listed in

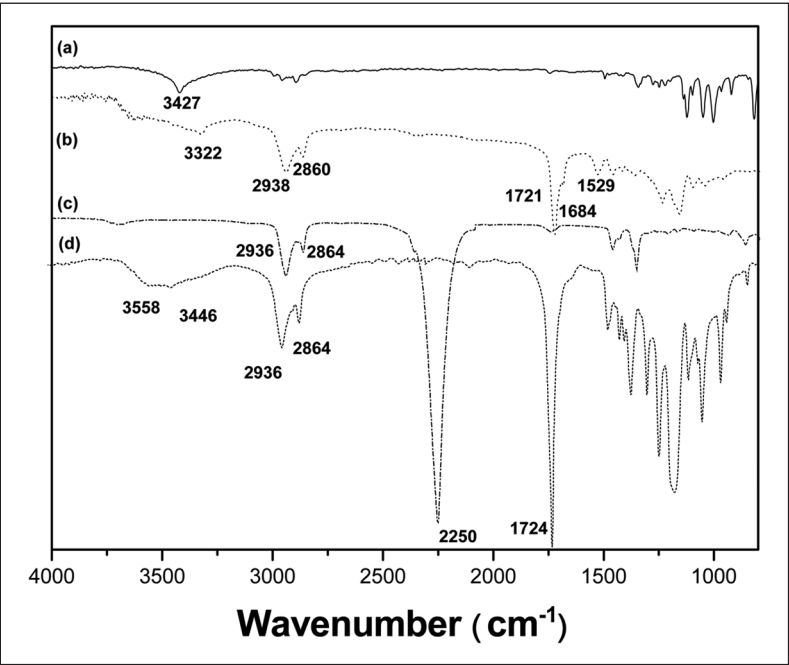


Figure 2. Fourier transform infrared (FT-IR) spectra of (a) isomannide, (b) polyurethane, (c) HDI, and (d) PCL diol 2000.
HDI: hexamethylene diisocyanate; PCL: poly(ϵ -caprolactone).

Table 2. Molecular weights, thermal properties, and surface energy of the synthesized polyurethanes.

	M_w^a (kDa)/mol	M_n^b (kDa)/mol	PDI ^c	T_g (°C) ^d	Water (°) ^e	Surface energy (mN m ⁻¹)
PU 1	145.0	101.8	1.4257	32.00 \pm 0.32	61.34 \pm 1.48	49.19 \pm 1.50
PU 2	68.2	44.6	1.5284	30.90 \pm 0.21	60.67 \pm 0.84	49.71 \pm 1.13
PU 3	30.8	20.2	1.5242	26.43 \pm 0.45	60.44 \pm 0.63	50.10 \pm 0.64
PU 4	16.9	12.6	1.2960	25.30 \pm 0.22	58.00 \pm 0.52	52.59 \pm 0.53

PDI: polydispersity index.

^aThe weight average molecular weights.

^bThe number average molecular weights.

^cPDI = M_w/M_n .

^dGlass transition temperature.

^eWater contact angle.

Table 2. Only one T_g was detected for all samples. The values of T_g of PU 1, PU 2, PU 3, and PU 4 were 32°C, 30.9°C, 26.4°C, and 25.3°C, respectively. A melting peak was not observed for any of the polymers examined.

Mechanical properties

The ultimate tensile strength (UTS) and Young's modulus for the polyurethanes are shown in Figure 3. The higher molecular weight polyurethane was obtained with PU 1, and this produced a

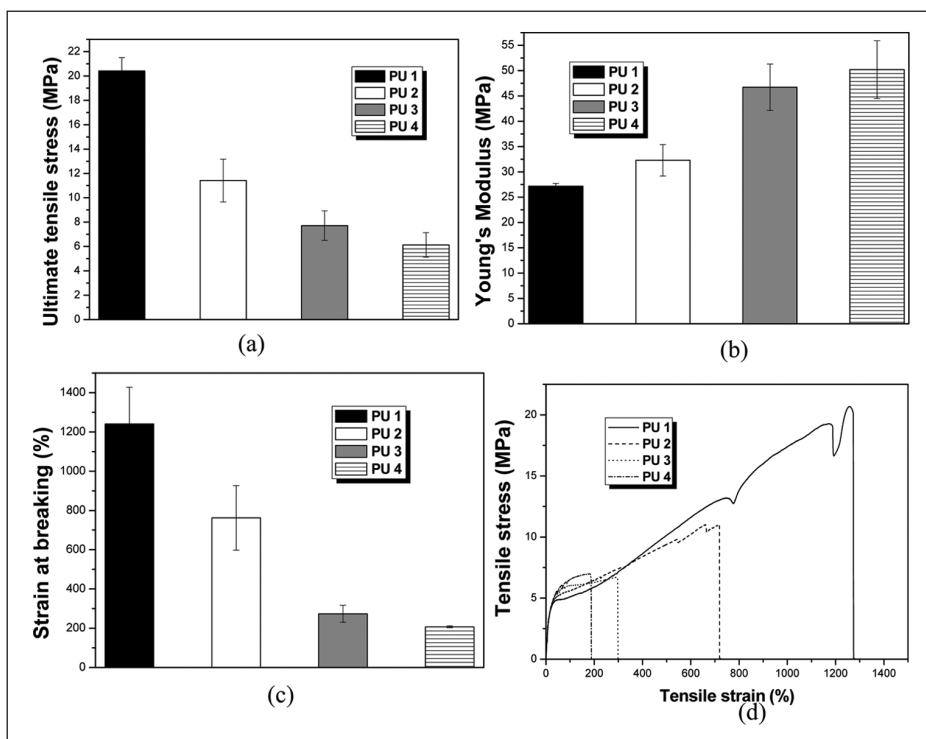


Figure 3. (a) Ultimate tensile strength (UTS), (b) Young's modulus, (c) strain at breaking (%), and (d) example of tensile stress–strain curves for the polyurethane series.

mean stiffness (called Young's modulus) of 27.19 ± 0.5 MPa, which was similar to that of ethyl–methyl acrylate. The UTS was 20.4 ± 1.09 MPa. Lower molecular weight polyurethane was obtained with PU 4, and this produced a mean stiffness of 50.19 ± 5.7 MPa and a UTS of 6.12 ± 1.00 MPa.

In vitro degradation test

The degradation characteristics of the polyurethane series were studied *in vitro* using a phosphate buffer solution at 37°C. Degradation profiles of the polyurethanes are presented in Figure 4 with the percentage weight loss of the polymers with time. It is evident that the polyurethane series was undergoing slow degradation. The changes in the polyurethane film surface after degradation (Figure 5) were determined by scanning electron microscope (SEM) (model JSM-6510; JEOL, Tokyo, Japan) photographs. All surfaces appear relatively smooth initially with a few defects. However, all of the polyurethanes had ~7% weight lost over 2 weeks. A continual slow weight loss of ~12% was observed for all of polyurethanes between weeks 2 and 8.

CCK-8 assay

The results of CCK-8 absorbance values for MC3T3-E1 cells (CCK-8) adhesion to and proliferation on the polyurethane films compared to that of PCL are shown in Figure 6. The CCK-8 assay

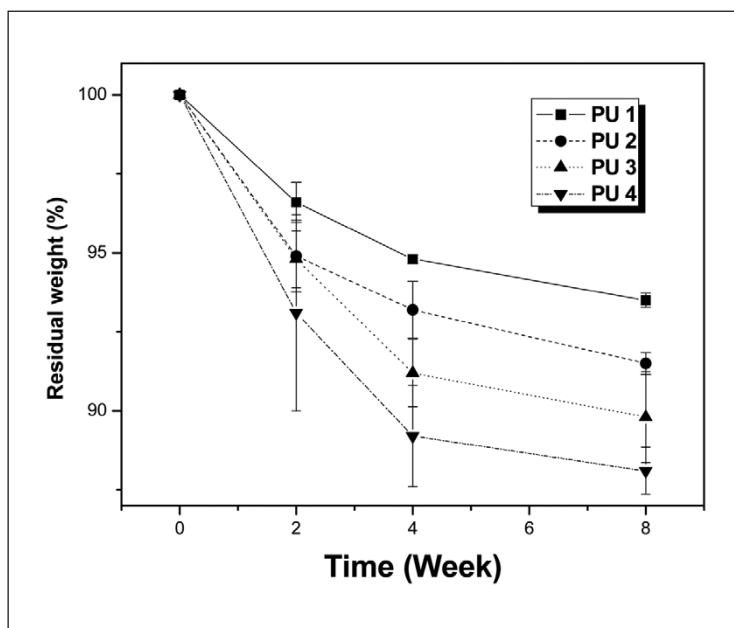


Figure 4. Residual weight (%) of the polyurethane films after degradation in phosphate buffer solution: (a) PU 1, (b) PU 2, (c) PU 3, and PU 4.

was implemented to measure mesenchymal stem cell (MSC) growth, and as it is specific to viable cells, it reflects not only the ability of the polyurethane film to promote cell adhesion and proliferation, but also its cytotoxicity. In comparison with PCL ($M_w = 60,000$, Sigma–Aldrich), relative cell numbers on the polymer films between days 1 and 7 of culturing on polyurethane films proliferated at least as well as those on PCL. Lower cell viability was observed from 1 to 7 days on the polyurethane surfaces in comparison with that of the control.

Cell attachment

Attachment of cells to a material is one of the prerequisites for evaluating biocompatibility for possible utilization in biomedical applications. The morphology of MC3T3-E1 cells after they were cultured on the polyurethane polymer films for 1–7 days was examined by SEM. The morphology of MC3T3-E1 cells attached to the wells were exhibited in Figure 7. The polyurethane films provided a suitable environment for cell attachment and spreading. The SEM images of the MC3T3-E1 cell morphology after being cultured on polyurethane film during 7 days PU 1, PU 2, PU 3, and PU 4 are shown in Figure 7(a) to (d), respectively. The cell culture results indicate that these polyurethanes are biocompatible and nontoxic.

Discussion

A great deal of research over the last decade has been conducted to develop biodegradable polyurethanes for medical applications.^{1,2} It would be advantageous for some applications if implants were produced from biodegradable and/or bioresorbable materials. Continuous degradation of the

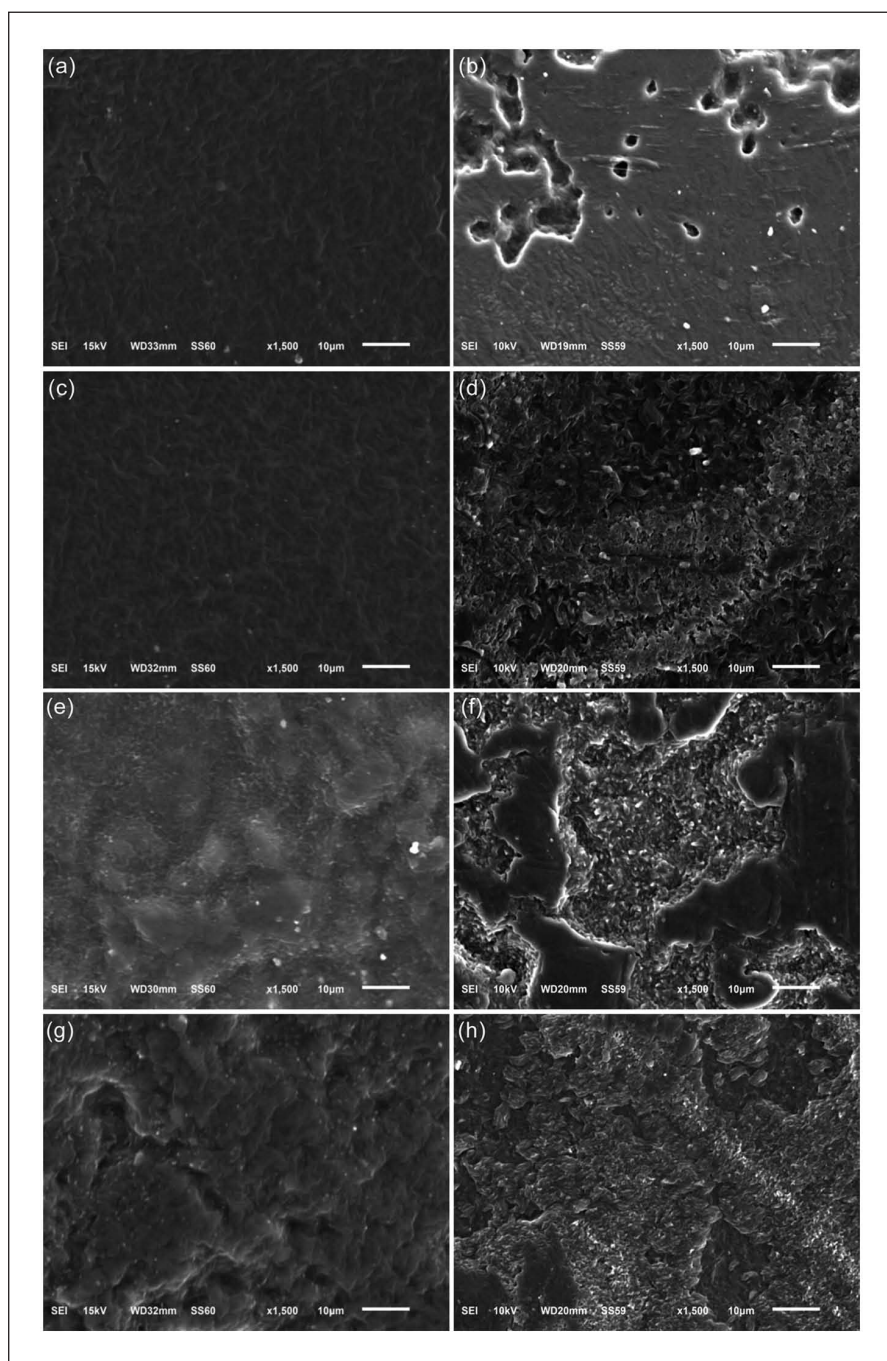


Figure 5. (a and b) SEM of PU 1 prior to degradation and at 2 months, (c and d) SEM of PU 2 prior to degradation and at 2 months, (e and f) SEM of PU 3 prior to degradation and at 2 months, and (g and h) SEM of PU 4 prior to degradation and at 2 months. SEM: scanning electron microscope.

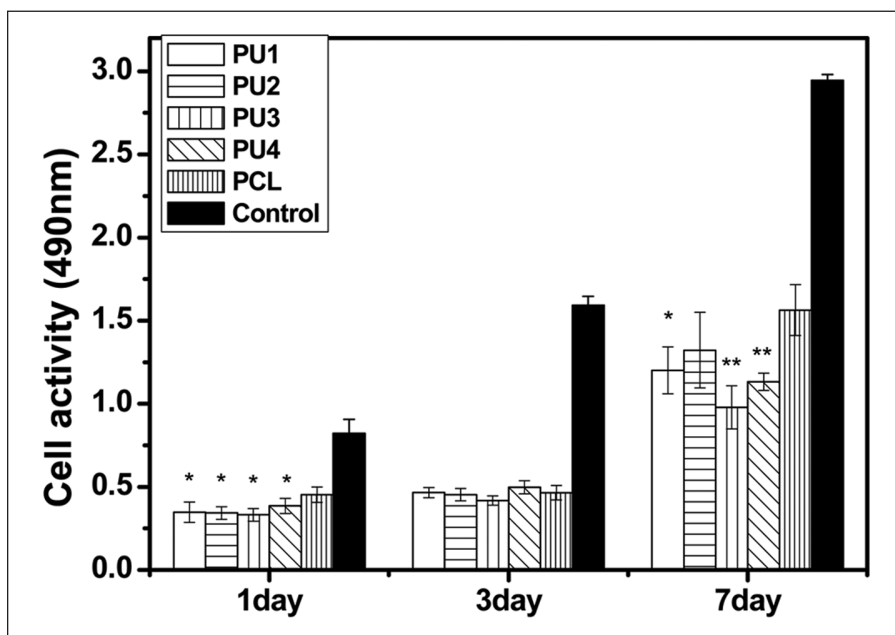


Figure 6. CCK-8 assay of rat MC3T3-E1 cultured on polyurethane wells showed more metabolically active cells on polyurethanes during the first 7 days.

ANOVA: analysis of variance; CCK: cell counting kit; PCL: poly(ϵ -caprolactone).

Data are mean \pm standard deviation. Significantly different from the control (* $p < 0.05$, ** $p < 0.01$, ANOVA, $n = 3$). Normalized values are shown.

implant material may then be accompanied by the formation of new organs and/or tissues.¹⁶ Biodegradable polyurethane elastomers are promising materials for such implants.¹⁷

The goal of this study was to evaluate the cytocompatibility of experimental polyurethanes with different isomannide-to-PCL diol content ratios in MC3T3-E1 cells. Our one-pot polymerization process is very simple and reduces the risk of toxicity from a surfactant or catalyst. In the molecular weight study, we should consider the reactivity of isomannide. Isomannide may have lower reactivity. The effect of the lower reactivity secondary hydroxyl of isomannide appeared to be quite evident with PU 1, which had the lowest isomannide content, and showed the highest M_w . This was also supported by the polymer yield data. The variations in the T_g of the soft segment as a function of composition have been suggested as an indicator of the degree of microphase separation in thermoplastic polyurethane elastomers.¹⁸ In general, the T_g value of soft segments decreases with increasing soft-segment molecular weight and decreasing diisocyanate content, although the changes are quite small. The T_g value also increased with increasing molecular weight.¹⁹ Factors with an effect on T_g in polyurethanes based on aliphatic diisocyanate include crystallization of the soft-segment and hard-segment components, the steric hindrance of the hard-segment unit during hydrogen bonding, and the inherent solubility of the hard and soft components.²⁰ If the T_g value is above body temperature, material will be in a rigid state.²¹ On the other hand, if the T_g of the polymer is below body temperature, the material would be in a rubbery state. For the polyurethanes in this study, the T_g decreased with decreasing molecular weight and T_g increased with decreasing contents of isomannide as expected. All four polyurethanes had a T_g below body temperature; thus, these polyurethanes could exhibit elastomeric properties at body temperature. Elastomeric

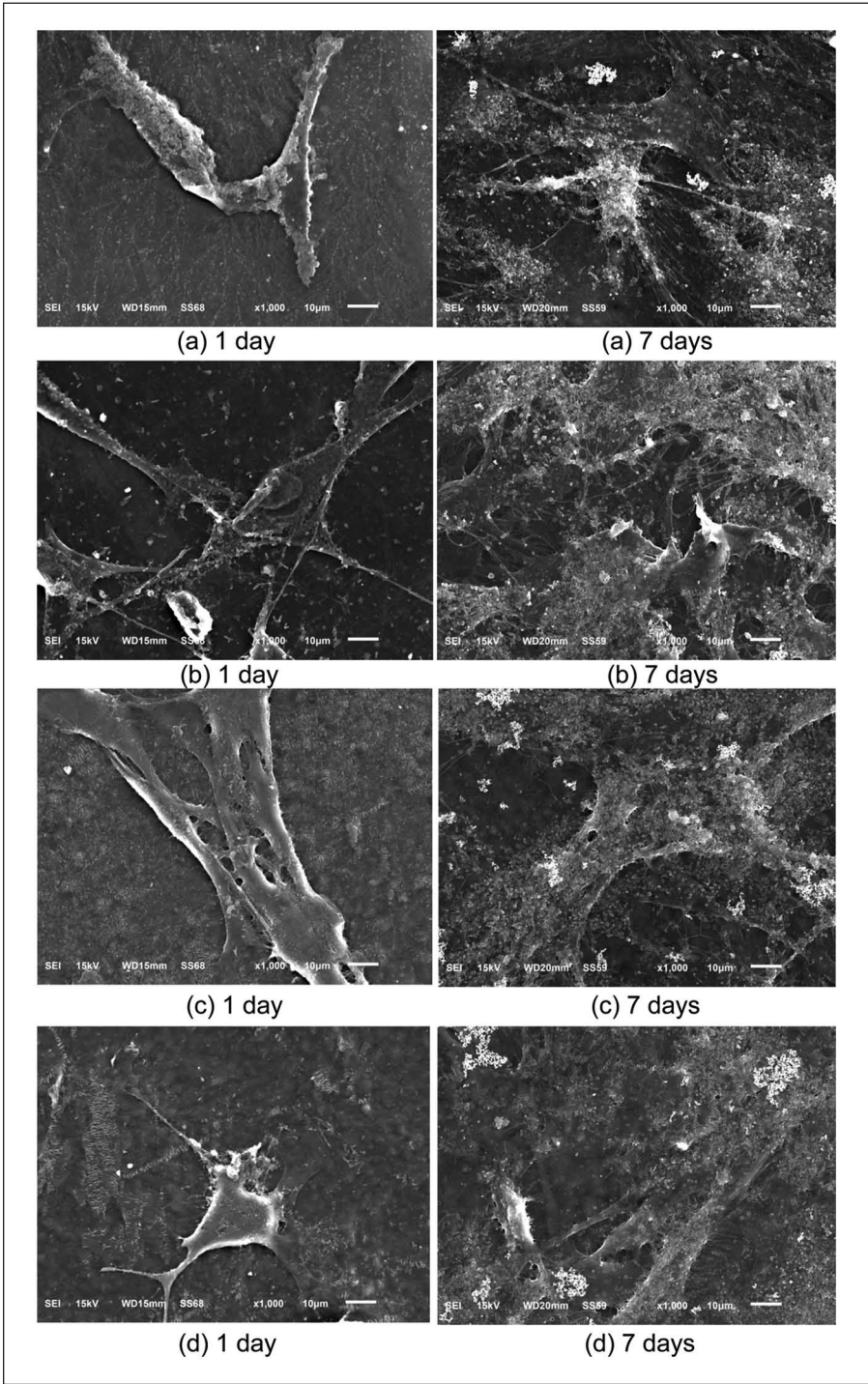


Figure 7. SEM images illustrating morphology of MC3T3-E1 cells after being cultured on polyurethane film during 7 days (a) PU 1, (b) PU 2, (c) PU 3, and (d) PU 4. SEM: scanning electron microscope.

properties at body temperature are an important quality for polymers in soft tissue engineering.²¹ The polyurethane films were flexible with tensile strengths from 6.12 ± 1.0 to 20.40 ± 1.09 MPa, stiffness from 27.2 ± 0.51 to 50.19 ± 5.7 MPa, and breaking strains from 208% to 1239%. The UTS of the films was comparable to that of the aorta (50%–100%),²¹ whereas the breaking strains were generally greater than that of the aorta. The mechanical properties of the film depended on the molecular weight, hard segments, and soft segments. As the molecular weight of a polymer increases, tensile strength increased. In addition, there was a relationship between hard segment and stiffness.²² The polyurethane stiffness increased with increasing isomannide (hard segment) levels. The main degradable polymers used in biomedicine are PCL, polyglycolic acid (PGA), and poly(L-lactic acid) (PLLA) and their derivatives. These polymers are all very stiff and are not suitable for utilization for tissue engineering of soft tissues, even in a fibrous or woven state. The development of softer, more flexible, and degradable polymers, as shown in this study, opens up a range of opportunities for soft tissue augmentation and regeneration, and the polymer family synthesized in this study offers such properties.

Much attention has been paid to the surface properties and structure of polymers, because the functional groups on the surface play an important role in the interactions of the materials with cells.²³ Contact angles have been used to estimate surface properties. The measured contact angles are listed in Table 2, and all of the polyurethane water contact angles were $<61^\circ$. It is generally agreed that hydrophilic surfaces have a contact angle with water in the range of 1° – 30° , whereas hydrophobic surfaces have contact angles of $>90^\circ$. Therefore, our polyurethanes had values somewhere between these ranges. As the isomannide content, which is a hydrophilic unit monomer, increased, the water contact angle of the polyurethanes decreased, indicating that the hydrophilicity of the polyurethanes can be controlled, although the changes were relatively small.

The rate of degradation of polyurethanes in aqueous media by hydrolysis is controlled mainly by the presence of urethane and ester bonds and varies significantly with the polymer chemical structure.²⁴ The acidic carboxyl group accelerates the hydrolysis of the polyester segments.¹⁸ The biodegradability of polyurethanes also depends on the molecular weight, degree of crystallinity, chemical structure, as well as susceptibility to microbial attack.²⁶ The number of carbons in the dialcohol seems to play an important role during hydrolytic degradation. The isomannide gives higher hydrophilicity to the polyurethane segment than does PCL diol, making hydrolytic attack easier. The rate of hydrophilicity and hard-segment contents had a more dominant effect on biodegradation rate rather than crystallinity for the polyurethanes presented here.²⁵

Many studies have ascertained biocompatibility using the CCK-8 or 3-(4,5-dimethylthiazol-2-yl)-5-(3-carboxymethoxyphenyl)-2-(4-sulfophenyl)-2H-tetrazolium (MTS) assay.¹⁸ When cell proliferation on a polymer is maintained for 1–2 weeks,²⁶ then the polymer can be considered cytocompatible and biocompatible. MC3T3-E1 cells were used as they are highly sensitive to culture conditions. The cell number was assessed via the CCK-8 assay due to its reliability and sensitivity.²⁷ The cells showed lower numbers from 1 to 7 days compared with the control, when seeded on polyurethane surfaces (Figure 7); however, the cells began to proliferate and increase in number. In comparison with PCL, relative cell number on polyurethane films in 7 days of culture was approximately the same as PCL. Although the cells showed a lower rate of proliferation than that of the control, the polyurethane films were considered to be cytocompatible. Furthermore, the synthesized polyurethanes had good mechanical strength, thermal stability, cytocompatibility, and biodegradability.

Conclusion

Polyurethane biobased isomannide polymers were synthesized from HDI via a simple one-pot polymerization. The important advantage was not to require a catalyst, solvent, or surfactant during

the reaction, which offered clinical advantages related to toxicity. Based on the in vitro mouse osteoblast cultures, the cells were suppressed slightly compared to the control but they grew well on the polyurethane films for 5 days. The CCK-8 cytotoxicity assay indicated that the polyurethanes had good biocompatibility. Taking into account the biocompatibility, degradability, and mechanical properties, these biobased polyurethanes have potential for soft tissue reconstruction, such as bladder, tracheal, and cardiovascular applications.

Declaration of conflicting interests

The authors declare that there is no conflict of interest.

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